

Succinyl-CoA synthetase and succinyl-CoA:3-oxoacid CoA transferase

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Introduction

Succinyl-CoA synthetase (SCS) is responsible for carrying out two unrelated but vital metabolic functions. One, it catalyzes the substrate-level phosphorylation step of the citric acid cycle [1], and two, it replenishes succinyl-CoA for ketone body catabolism [2] and for porphyrin synthesis [3]. The reversible reaction catalyzed by SCS,

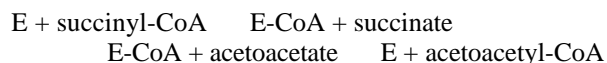


uses either adenosine or guanosine nucleotides. Although some species have a form of SCS that utilizes both ATP and GTP, other species have isoforms of SCS specific for each nucleotide [4]. The nucleotide-specific isoforms are believed necessary for SCS to serve its different metabolic roles. Thus, the GDP-specific SCS could maintain the high level of succinyl-CoA required for ketone body metabolism [5] (i.e., the reverse reaction above), since the mitochondrial GTP:GDP ratio is approximately 100 [6]. In contrast, since the mitochondrial ATP:ADP ratio is approximately 1, ADP-specific SCS would be suitable for the role played in the citric acid cycle [7].

E. coli SCS, in the presence of CoA, crystallizes in the tetragonal space group $P4_322$ with unit cell dimensions $a = b = 98.68 \text{ \AA}$, $c = 403.76 \text{ \AA}$, and $\alpha = \beta = \gamma = 90^\circ$ [8]. We have solved the structure [9] and refined it at 2.3 \AA resolution using data from 23 crystals [10]. The binding site for ADP was determined using crystals soaked with ADP and Mg^{2+} ions, proving that the histidine residue phosphorylated in the reaction must swing approximately 35 \AA between the binding site for CoA and the binding site for nucleotide [11]. To understand the catalytic mechanism, the binding site of succinate or the succinyl moiety must be known. Crystals of pig heart GTP-specific SCS grow in the absence of CoA [12]; we have successfully soaked them with CoA to determine the structure of the complex. The goals of our work at the Advanced Photon Source were 1) to determine the structure of *E. coli* SCS soaked with GDP and Mg^{2+} ions to prove that the guanosine nucleotide binds in the same site as the adenosine nucleotide and define the determinants of nucleotide specificity, 2) to collect a full data set from a single crystal of *E. coli* SCS at cryotemperatures, and 3) to determine the structure of pig heart GTP-specific SCS soaked with a nonhydrolyzable succinyl-CoA analogue to define the succinyl-binding site.

Succinyl-CoA:3-oxoacid CoA transferase (SCOT) is required for the catabolism of ketone bodies that are produced primarily in the liver as a source of metabolic fuel for other tissues. In the catabolism of ketone bodies, SCOT activates acetoacetate by transferring CoA from succinyl-CoA to

produce acetoacetyl-CoA and succinate. The reaction proceeds with the participation of a glutamate residue of the enzyme in the formation of a thioester linkage to CoA:



We have crystallized SCOT in a monoclinic crystal form, space group $P2_1$ with unit cell dimensions $a = 62.97 \text{ \AA}$, $b = 262.01 \text{ \AA}$, $c = 60.81 \text{ \AA}$, $\alpha = \beta = 90^\circ$, and $\gamma = 112.31^\circ$, but the structure determination by multiple isomorphous replacement methods has been hampered by nonisomorphism. We found a second crystal form, $P2_12_12$, with unit cell dimensions $a = 101.42 \text{ \AA}$, $b = 142.94 \text{ \AA}$, $c = 69.32 \text{ \AA}$, and $\alpha = \beta = \gamma = 90^\circ$ that we hoped would provide a heavy-atom derivative useful for phasing.

Methods and Materials

Crystals of *E. coli* and pig heart GTP-specific SCS were grown as described in references [9] and [12]. Crystals of *E. coli* SCS were soaked with GDP and Mg^{2+} ions using the protocol worked out for ADP [11]. Crystals of pig heart GTP-specific SCS were soaked with the nonhydrolyzable succinyl-CoA analogue, S-butaryl-CoA [13]. We grew the monoclinic and the orthorhombic crystals of SCOT as well as crystals of a deletion mutant in which five residues (thought to be important in the determination of the quaternary structure of SCOT) had been removed [14]. The data were collected at the BioCARS beamline, 14-BM-C.

Results

Data from crystals of *E. coli* SCS were collected with a crystal-to-detector distance of 250 mm. The crystal soaked with GDP and Mg^{2+} ions diffracted to 2.7 \AA , with an overall R_{merge}^\dagger of 6.2% and an R_{merge} of 36% on data in the resolution range between 2.8 and 2.7 \AA .

Data from a crystal of pig heart GTP-specific SCS soaked with S-butaryl CoA were collected with a crystal-to-detector distance of 200 mm. This resulted in a data set complete to 2.3 \AA , but only 85.4% complete to 1.95 \AA resolution. R_{merge} was 4.9% overall and 25% for data in the resolution range from 2.02 to 1.95 \AA .

Two "native" data sets were collected for SCOT (one from a crystal in the monoclinic crystal form and the second from the deletion mutant in the orthorhombic crystal form). Data

[†] $R_{\text{merge}} = \left(\frac{\sum (I_i - \bar{I})}{\sum I_i} \right)$, where I_i is the intensity of an individual measurement of a reflection and \bar{I} is the mean value for all equivalent measurements of this reflection.

from the deletion mutant processed to 2.1 Å gave an overall R_{merge} of 3.7%. In the resolution range from 2.18 to 2.1 Å, R_{merge} was 8.7%. This data set was 92.3% complete overall, but only 60% complete in the resolution range between 2.2 and 2.1 Å. Surprisingly, the orthorhombic crystal of the deletion mutant was not isomorphous with the orthorhombic crystals of wild-type SCOT. A self-rotation map indicated that the noncrystallographic two-fold axis is 15° from the b-axis of the crystals for the deletion mutant, while it superimposed with one of the crystallographic two-fold axes (presumably the b-axis) in crystals of wild-type SCOT. Data were also collected from six heavy atom soaks of orthorhombic crystals of SCOT. These data sets were processed to 2.5 Å. The most promising is from a crystal soaked with iridium, but the crystal displayed high mosaicity and the electron density maps calculated with phases from the derivative data do not show a clear boundary between the protein and the solvent. Work on phasing the SCOT structure is continuing.

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References

- [1] S. Kaufman, C. Gilvarg, O. Cori, and S. Ochoa, *J. Biol. Chem.* **203**, 869–888 (1953).
- [2] J.A. McClellan, and J.H. Ottaway, *Comp. Biochem. Physiol.* **67B**, 679–684 (1980).
- [3] R.F. Labbe, T. Kurumada, and J. Onisawa, *Biochim. Biophys. Acta* **111**, 403–415 (1965).
- [4] P.D. Weitzman, T. Jenkins, A.J. Else, and R.A. Holt, *FEBS Lett.* **199**, 57–60 (1986).
- [5] J.H. Ottaway, J.A. McClellan, and C.L. Saunderson, *Int. J. Biochem.* **13**, 401–410 (1981).
- [6] C.M. Smith, J. Bryla, and J.R. Williamson, *J. Biol. Chem.* **249**, 1497–1505 (1974).
- [7] T.M. Jenkins, R. Eisenthal, and P.D.J. Weitzman, *Biochem. Biophys. Res. Commun.* **151**, 257–261 (1988).
- [8] W.T. Wolodko, M.N.G. James, and W.A. Bridger, *J. Biol. Chem.* **259**, 5316–5320 (1984).
- [9] W.T. Wolodko, M.E. Fraser, M.N.G. James, and W.A. Bridger, *J. Biol. Chem.* **269**, 10883–10890 (1994).
- [10] M.E. Fraser, M.N.G. James, W.A. Bridger, and W.T. Wolodko, *J. Mol. Biol.* **285**, 1633–1653 (1999).
- [11] M.A. Joyce, M.E. Fraser, M.N.G. James, W.A. Bridger, and W.T. Wolodko, *Biochemistry* **39**, 17–25 (2000).
- [12] M.E. Fraser, M.N.G. James, W.A. Bridger, and W.T.

Wolodko, submitted to *J. Mol. Biol.*

- [13] D.P. Martin and D.G. Drueckhammer, *J. Am. Chem. Soc.* **114**, 7287–7288 (1992).
- [14] J.-C. Rochet, E.R. Brownie, K. Oikawa, L.D. Hicks, M.E. Fraser, M.N.G. James, C.M. Kay, W.A. Bridger, and W.T. Wolodko, submitted to *Biochemistry*.