

# The Structure of the Manganese Oxide on the Cell Wall of the Bacterium *Leptothrix discophora*: An XAFS Study

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## Introduction

In natural waters, manganese oxides ( $\text{MnO}_x$ ) are important in mediating the bioavailability of trace metals such as Ni, Cd, and Pb, because these metals readily adsorb to the  $\text{MnO}_x$  surface. In the presence of oxygen and under neutral ( $\text{pH} = 7$ ) conditions, soluble manganese ( $\text{Mn}^{2+}$ ) is oxidized to  $\text{MnO}_x$ . Bacteria such as *Leptothrix discophora* can catalyze this process.

In our ongoing studies<sup>1</sup> of trace metal bioavailability, we grew *L. discophora* in the presence of  $\text{Mn}^{2+}$  and employed Mn K-edge x-ray absorption spectroscopy to determine the oxidation state (XANES) and coordination (EXAFS) of Mn atoms in the bacterial  $\text{MnO}_x$ . Our results suggest that this  $\text{MnO}_x$  has a structure closely related to the mineral birnessite.<sup>2</sup>

## Methods and Materials

*L. discophora* SP-6 was grown for 48 h under controlled laboratory conditions at room temperature.<sup>2</sup> All fluid was removed, and the bacteria were freeze-dried. Mn K-edge absorption spectra were collected using the Si-(111) double crystal monochromator on the Pacific Northwest Consortium Collaborative Access Team undulator beamline.<sup>3</sup> The sample was sandwiched between two layers of adhesive Kapton tape, mounted along the circumference of a Plexiglas disk (3.0"  $\varnothing$ ), and rotated at 1200 rpm to minimize radiation damage. The fluorescence spectra were collected using an ion chamber.<sup>4</sup> Spectra of manganese oxide minerals were measured under the same experimental conditions to aid in the structural identification.

EXAFS oscillations were extracted from the spectra using our standard procedures.<sup>2</sup> Fourier transformation into r-space of the  $k^3$ -weighted EXAFS spectrum  $k^3\chi(k)$  was performed over the  $k$ -range from 2.9 to 11.85  $\text{\AA}^{-1}$  using a 10% Gaussian window. The pre-edge features were isolated from the absorption edge using a background subtraction procedure.<sup>2</sup>

Table 1: The experimental Mn K-edge absorption energies

Name	Oxidation state of Mn	Edge energy (eV) <sup>(a)</sup>	White-line peak energy (eV) <sup>(a)</sup>
Mn metal foil	0	6537.7 <sup>(b)</sup>	
$\text{MnO}_2$ (birnessite)	3.77	6548.0	6559.7
<i>Leptothrix discophora</i>	4	6548.4	6560.5
$\text{MnO}_2$ (todorokite)	4	6549.5	6560.9
$\text{MnO}_2$ (powder)	4	6551.4	6559.0

(a) Uncertainty:  $\pm 0.5$  eV. (b) Calibration standard.<sup>9</sup>

## Results and Discussion

The edge energy and the white-line energy for the bacterial  $\text{MnO}_x$  are between those of birnessite and todorokite (Table 1), suggesting a formal oxidation state close to 4.0 ( $\text{Mn}^{4+}$ ). The spectral features resemble those of birnessite and todorokite, but there is a shoulder preceding the white line observed in neither mineral (Fig. 1). There is one low-intensity pre-edge feature peaking at

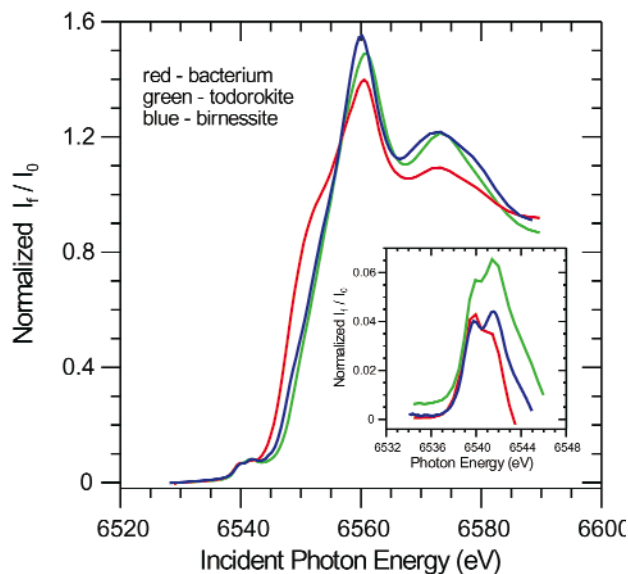


FIG. 1. Mn K-edge XANES spectra.

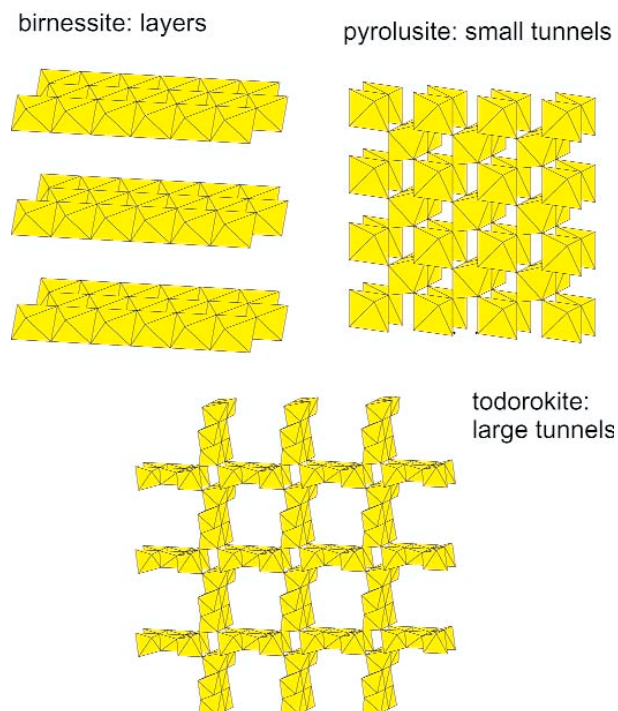


FIG. 2. Some  $\text{MnO}_2$  crystal structures.

6539.6 eV with a high-energy shoulder caused by a peak at 6541.5 eV. The relative intensities of these peaks are reversed in both minerals (Fig. 1, inset).

$\text{MnO}_2$  has a variety of different stable crystal structures,<sup>5,6</sup> composed of  $\text{MnO}_6$  octahedra forming layers or tunnels (Fig. 2). The Fourier transform of  $k^3\chi(k)$  of all manganese dioxides has

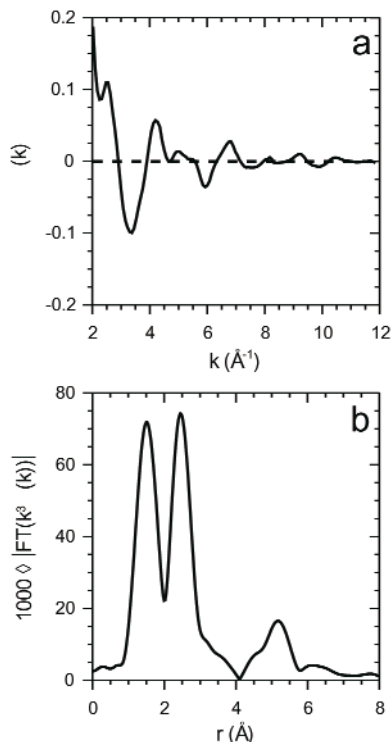


FIG. 3. (a) Mn K-edge XAFS spectrum of the bacterium and (b) its Fourier transform.

two fairly high intensity peaks corresponding to the Mn-O distance of 1.9 Å in the first coordination shell and the edge-sharing Mn-Mn distance of 2.9 Å in the second shell. A peak at 3.4 Å, corresponding to the Mn-Mn distance of corner-sharing octahedra, is only observed for the tunnel structures.<sup>5,7</sup>

The Fourier transform of the EXAFS interference function  $\chi(k)$  [Fig. 3(a)], weighted by  $k^3$  of the bacteria sample has two high-intensity peaks corresponding to the Mn-O distance of 1.9 Å and the Mn-Mn distance of 2.9 Å [Fig. 3(b)]. A peak at 3.4 Å is not observed, indicating a layered structure. The peaks in the figure appear at shorter distances because of the EXAFS phase shift, which has not been removed. Fits using spectra calculated with

FEFF7<sup>8</sup> suggest that the bacterial MnO<sub>x</sub> is composed of microcrystals having a structure related to birnessite.<sup>2</sup>

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## References

- <sup>1</sup> A. Jürgensen, J.R.M. King, R.A. Gordon, D.-T. Jiang, L.I. Bendell-Young, M. Moore, and E.D. Crozier, Advanced Photon Source User Activity Report, **ANL/005**, (Argonne National Laboratory, Argonne, IL, 2000) 434-435 (2000).
- <sup>2</sup> A. Jürgensen, J.R.M. King, R.A. Gordon, L.I. Bendell-Young, M.M. Moore, and E.D. Crozier, submitted to *Geochim. Cosmochim. Acta* (2001).
- <sup>3</sup> S.M. Heald, D.L. Brewster, E.A. Stern, K.H. Kim, F.C. Brown, D.T. Jiang, E.D. Crozier, and R.A. Gordon, *J. Synchrotron Radiat.* **6**, (3), 347-349 (1999).
- <sup>4</sup> D.T. Jiang and E.D. Crozier, *Can. J. Phys.* **76**, 621-643 (1998).
- <sup>5</sup> A. Manceau and J. M. Combes, *Phys. Chem. Miner.* **15**, 283-295 (1988).
- <sup>6</sup> J.-B. Li, K. Koumoto and H. Yanagida, *J. Mater. Sci.* **23**, 2595-2600 (1988).
- <sup>7</sup> Alain Manceau, A. I. Gorshkov, and V. A. Drits, *Am. Mineral.* **77** 1144-57 (1992).
- <sup>8</sup> S.I. Zabinsky, J.J. Rehr, A. Ankudinov, R.C. Albers, and M.J. Eller, *Phys. Rev. B.* **52**, 2995-3009 (1995).
- <sup>9</sup> S. Kraft, J. Stümpel, P. Becker, and U. Kuetgens, *Rev. Sci. Instrum.* **67**, 681-687 (1996).